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Key Vocabulary:

coefficient,

combination,

compound,

decomposition,

double

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replacement,
element,
molecule,
product,
reactant, single
replacement,
subscript Prior
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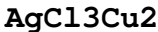
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prior knowledge
and get students
thinking.

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each of the real
chemical
formulas below,
calculate how

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many of each
element there
are. Check your
answers for the
first three
formulas using
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Therefore, a
balanced
chemical
equation will
show the same
number of each
type of atom on
each side of the

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- Name Arianna
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Introduction. In the process students will be required to balance chemical equation, exhibiting proficiency of the Science and Engineering Practice of Mathematics and Computational

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Thinking. During this lesson students will recognize that energy and matter is an underlying crosscutting concept involved in all chemical reactions.

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a reaction in
which an element

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reacts with a compound to form a new compound and a different element. a number in a chemical formula representing the number of atoms of a particular element in one molecule of the compound.

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a number that

multiplies a

term in an

equation.

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the number of
atoms on each
side is
presented as
visual,
histogram, and
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Balancing
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The equation is balanced when there are equal numbers of each type of atom represented on each side of the equation. In the Gizmo, use the up and down arrows to adjust the numbers of hydrogen, oxygen, and

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water molecules
until the
equation is
balanced. When
you are done,
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