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– 2 – Section A Answer all

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questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid, $\text{HOOC?COOH}\cdot x\text{H}_2\text{O}$, by titrating a known mass of the acid with a $0.100 \text{ mol dm}^{-3}$ solution of

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NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

Chemistry Standard level Paper 2 - Xtreme

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[2] (b) Marit arranged the values she
found in Table 12 into an energy
cycle. $C_2H_4(g) + H_2(g) \rightarrow C_2H_6(g)$ 3 0
2 $2CO_2(g) + 3H_2O$ - 1 4 1 1 k J m o l
- 1 - 2 8 6 k J m o l - 1 - 1 5 6 0 k J

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m o l – 1 (hydrogenation) (l)

Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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[2] (ii) Identify the oxidizing agent in step 1..... [1] (b) A student carried out this experiment three times, with three identical small brass nails, and obtained the following results. Mass of brass = $0.456 \text{ g} \pm 0.001 \text{ g}$ titre 1 2 3 initial volume of $0.100 \text{ mol dm}^{-3} \text{ S 2 o 3 2-}$ (± 0.05

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cm)³ 0.00 0.00 0.00 Final volume of
0.100 mol dm⁻³ S 2 o 3

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