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water molecules for the positive ion is stronger than the attraction of the negative ions near it in the crystal, the ion leaves the crystal and enters solution. After entering solution, the dissolved ion is surrounded completely by water molecules, which tends to prevent the ion from reentering the ...

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(b) With the information given in sample exercise 15.3 , determine the value of this equilibrium constant at 25 °C. b. a.

Writing products over reactants, we have  
SAMPLE EXERCISE 15.3 Interpreting  
the Magnitude of an Equilibrium Constant  
The reaction of N<sub>2</sub> with O<sub>2</sub> to form NO  
might be considered a means of “fixing” n  
itrogen:

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sulfur atom ( $A_r = 32$ ) weighs twice as much as an oxygen atom ( $A_r = 16$ ), 32 g of sulfur atoms and 16 g of oxygen atoms both contain the same number of atoms.

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plays an important role in the health, abundance and diversity of the aquatic life that can live in a stream. Excessive amounts of some constituents (nutrients), or the lack ... Most surface waters contain between 5 and 15 ppm DO. 2.

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The presentation of information in this chapter assumes that you can already perform the tasks ... Picture a layer of ethanol being carefully added to the top of some water (Figure 15.3). Because the particles of a liquid are moving constantly, some of the ethanol particles ... 576  
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