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In a non-flow process there is heat transfer loss of 1055 kJ and an internal energy increase of 210 kJ.

Determine the work transfer

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and state
whether the
process is an
expansion or
compression.

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The absolute
humidity ? is
the ratio of
masses of the
vapor and air,

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i.e., $\gamma = m v / m$

a. Now, applying

ideal gas

equation, $pV =$

mRT for water

vapor and for

air, we have,

since the volume

and temperature

are the same, γ

$= 0.622 p v / p$

a. The ratio of

specific gas

constants (R in

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preceding
equation) of
water vapor to
air equals 0.622

.

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Thermodynamics
is the study of
relationship
between energy
and entropy,
which deals with
heat and work.

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It is a set of theories that correlate macroscopic properties that we can measure (such as temperature, volume, and pressure) to energy and its capability to deliver work.

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Intensive and
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Properties
Concepts of
State, Process
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Basic

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1_Introduction & Basic Concepts

Thermodynamics
is the branch of
physics that
deals with heat
and temperature,
and their
relation to
energy, work,
radiation, and
properties of
matter. The

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behavior of
these quantities
is governed by
the four laws of
thermodynamics
which convey a
quantitative
description
using measurable
macroscopic
physical
quantities, but
may be explained
in terms of

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